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Effluent treatment process in molasses-based

Review

distillery industries: A review

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Abstract

Distillery effluent is a contaminated stream with high chemical oxygen demand (COD) varying from 45,000 to 75,000 mg/l and low pH values of between 4.3 and 5.3. Different processes covering aerobic, anaerobic as well vsico-chemic nethods which have been employed to this effluent has been given in this review paper. Among the different methods ailable, it was found that "An Inverse Anaerobic Fluidization" to be a better choice for treating effluent from molasses-based distillery inductive suing an interse anaerobic fluidized-bed reactor (IAFBR). This technology has been widely applied as an effective step in removing 80-854 of the COD in effluent stream. Therefore, in this review, attention has been paid to highlight in respect of fluidization phenomena, process per nance, stabil of the system, operating parameters, configuration of inverse anaerobic fluidization and suitable carrier material employed in an rse an obic fluidized-bed reactor especially for treating this effluent.

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Keywords: Molasses; Spentwash; Treatment Methods; Ana Quidized Leactor; Perlite; Inverse anaerobic fluidized-bed reactor

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1. Introduction

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In India, there are a number of large-scale distilleries integrated with sugar mills. The waste products from sugar mill comprise bagasse (residue from sugarcane crushing), pressmud (mud

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and dirt residue from juice clarification) and molasses (final residue from sugar crystallization). Bagasse is used in paper manufacturing and also as a fuel in boilers. Molasses is utilized as a raw material in distillery for ethanol production. Pressmud has no direct industrial application. Ethanol manufactured from molasses has large volumes of high strength wastewater that is of serious environmental concern. The effluent is characterized by extremely high COD (80,000-100,000 mg/l) and biochemical oxygen demand (BOD) (40,000-50,000 mg/l), apart from low pH, strong odor and dark brown color (Central Pollution Control Board (CPCB) 1994, 2003, India). India is the second largest producer of ethanol in Asia with a projected annual production of about 2300 million litres in 2006-2007 [1], alcohol distilleries have been rated as one of the 17 most polluting industries. Apart from high organic content, distillery wastewater also contains nutrients in the form of nitrogen (1660-4200 mg/l), phosphorous (225–3038 mg/l) and potassium (9600–17,475 mg/l) [2] that could lead to eutrophication of water bodies. Further, its dark color hinders photosynthesis by blocking sunlight and therefore deleterious to aquatic life [3]. The population equivalent of distillery wastewater based on BOD has been reported to be as high as 6.2 billion which means that the contribution of distillery waste in India to organic pollution is approximately seven times more than the entire Indian population. The wastewater from distilleries, major portion of which is spentwash, is nearly 15 times the total alcohol production. This massive quantity, approximately 40 billion liters of effluent, if disposed untrea would cause considerable stress on the watercourses leading to widespread damage to aquatic life. Adequate treatment therefore imperative before the effluent is disc In the ager present day context stringent statutory requirements hav the distilleries to improve the quality level to their eff made exploring viable alternatives. For instance, In distilleries were stipulated to achieve zero discharge of spentwhy to inland surface water by December 2005

Production of ethyl alcohol in distilleric based on care sugar molasses constitutes a major industry in Ask and South America. The world's total projection of alcohol from cane molasses is more than 13 million ³/anny. The aqueous distillery effluent stream known as spe values a dark brown, highly organic ximal 12–15 mes by volume of the effluent and is e ost complex, troublesome one of product alco л. It i and strong of organig ndustrial exquents, having extremely high COD and D: se of the high concentration of organic load, dlery spentwash is a potential source of renewable energy.

The wastewater from distilleries, major portion of which is spentwash, is nearly 15 times the total alcohol production. This massive quantity, approximately 40 billion liters of effluent, if disposed untreated could cause considerable stress on the watercourses leading to widespread damage to aquatic life.

In an earlier review on this subject [5], discussed treatment options practised in the 1970s and more recently [6], have described the characteristics and anaerobic treatment of effluent obtained from different feedstock used for ethanol production. The wastewater characterization and salient features of the existing treatment options and the recent advancement in the field of an inverse anaerobic fluidization [7] ave been discussed in the following sections.

2. Wastewater charace cization

the maje vastewat streams generated at Table 1 lists o the alcohol . different stage w .cturing process. Table 2 summarizes le typi Characteristers of spentwash generated in Indian distilleries using ugarcane molasses. Table 3 describes the Islandards of disting a filternt. Values for beet molassesd effluent have been given for comparison. The main source ba of astewater generation was the distillation step wherein large hes of dark bown effluent (termed as spentwash, stillage, vol slop vinasse were generated in the temperature range of 71-81 °C₁, The characteristics of the spentwash depended raw material used [11]; also, it has been estimated that 0 8% of me molasses constituents end up as waste [12]. Molasses spentwash has had very high levels of BOD, COD, COD/BOD ratio as well as high potassium, phosphorus and sulfate content (Table 2). In addition, cane molasses spentwash contains low molecular weight compounds such as lactic acid, glycerol, ethanol and acetic acid [6]. Cane molasses also contained around 2% of a dark brown pigment called melanoidins that imparted color to the spentwash [13]. This reaction proceeded effectively at temperatures above 50 °C and pH 4-7. The structure of melanoidins was still not well known [14]. Only 6-7% degradation of the melanoidins has been achieved in the conventional anaerobic-aerobic effluent treatment process [15]. Due to their antioxidant properties, melanoidins were toxic to many microorganisms involved in wastewater treatment [16]. Apart from melanoidins, spentwash contained other colorants such as phenolics, caramel and melanin. Phenolics have been more pronounced in cane molasses wastewater whereas melanin was significant in beet molasses [17].

Table 1

Sample quantities and characteristics of wastewater streams generated in an Indian distillery (S. Majumdar, personal communication)

Parameter	Specific wastewater generation (kl/kl alcohol)	Color	pН	Suspended solids (mg/l)	BOD (mg/l)	COD (mg/l)
Spentwash	14.4	Dark brown	4.6	615	36,500	82,080
Fermenter cleaning	0.6	Yellow	3.5	3000	4000	16,500
Fermenter cooling	0.4	Colorless	6.3	220	105	750
Condenser cooling	2.88	Colorless	9.2	400	45	425
Floor wash	0.8	Colorless	7.3	175	100	200
Bottling plant	14	Hazy	7.6	150	10	250
Other	0.8	Pale yellow	8.1	100	30	250

Table 2	
Characteristics of spentwash generated from various feedstock	[18,6,2]

Characteristics	Feedstock				
	Cane molasses	Beet molasses			
	Pathade [18]	Mahimairaja and Bolan [2]	Wilkie et al. [6]		
COD (mg/l)	65,000–130,000	104,000–134,400	91,100		
BOD (mg/l)	30,000-70,000	46,100–96,000	44,900		
COD/BOD ratio	2.49		1.95		
Total solids (mg/l)	30,000-100,000				
Total suspended solids (mg/l)	350				
Total dissolved solids (mg/l)	80,000	79,000–87,990			
Total nitrogen (mg/l)	1000-2000	1660–4200	3569		
Total phosphorus (mg/l)	800-1200	225-3038	163		
Potassium (mg/l)	8000-12,000	9600–17,475	10,030		
Sulfur as SO_4 (mg/l)	2000-6000	3240–3425	3716		
pH	3–5.4	3.9–4.3	5.35		

3. Treatment methods

There are various technologies for treatment of raw spentwash which utilize resource recovery and disposal. The secondary and tertiary treatments on their own are not technically and economically viable options for mitigating the problems associated with treatment and disposal of highstrength spentwash. Post-anaerobic effluent has had high COD, BOD and suspended solids (SS). Solar evaporation required a large land area and was non-functional during the rainy se [9]. Two-stage aerobic treatment has high-energy requiren nts. Partially treated spentwash has had a high (carbon: nitrogen) N ratio (>20). It could reduce the availability of imnt mine nutrients by trapping them into immobile anic t ms, an might produce phytotoxic substances during decom sition. It has been unsuitable for irrigation. Tertiary the mer physico-chemical methods, adsorption and ad ced chemical d for remo oxidation processes, basically ad of color in addition to trace organics, inversed h, operational st. Stringent regulations on the dimharge of comed effluent impeded direct discharge of secondary treated efflue Evaporation and incineration have not seen widdly adopted due to their being highly energy interve. Di dlery spentwash also contained nd organize content. Its use has been high nitrogen, phospho. agare te productivity; furthermore further repo ALL creas under corf.olled c Iffluent was capable of replacditions, ing appeation o nic fertilizers [19,20]. However, for the high stren. alasses-based spentwash, the odor, putrefaction andscape due to unsystematic disposal are conand unpleasa cerned in land a lication. In addition, this option was subject to land availability in the vicinity of the distillery; also, it was

be location in a low-medium rainessential that the Asposa. tigations have indicated that land fall area [5]. re recent in v effluent contamdisposal of Astra ination [21]. Deep U disposal was another option but limited nd storage a specific geological location limit this und ernative. Other disposal methods like evaporation of spentash to produce animal feed and incineration of spentwash for tash recovery ave also been practised [5,6]. Considering the lems in the treatment and disposal of distillery spentwash pì Ko surface water, waste management option with on la ourse to zero effluent discharge would appear to be more prom. Ing. Fig. 1 presents the options currently employed for molasses spentwash treatment.

3.1. Biological treatment methods

3.1.1. Aerobic process

Post-anaerobic effluent has had high BOD, COD and SS. Solar evaporation required a large land area and also needed to take into consideration the weather conditions prevailing in the region. It was also non-functional during the monsoon [9]. The other aerobic methods have been described below.

3.1.1.1. Biocomposting process. Biocomposting is a method of activated bioconversion through the aerobic route, whereby heterotrophic microorganisms act on carbonaceous materials depending on the availability of the organic source and the presence of inorganic material essential for their growth. Composting is particularly effective in converting the wet materials to a usable form thereby stabilizing the organics and destroy-

Table 3

ISI Standards (IS: 506-1980) for limits of composition of distillery effluent

Characteristics	For discharge into water course	For discharge into public sewers	For discharge on land	
PH	5.5–9.0	5.5–9.0	5.5–9.0	
BOD5 (mg/l)	100	500	500	
Total suspended solids (mg/l)	100	600	_	
Oil and grease (mg/l)	10	100	6–10	
Temperature °CShall not exceed 40 °C in any section of the stream within 15 m from effluent outlet				



Fig. 1. Spentwash treatment options.

ing the pathogenic organisms in addition to significant drying of the wet substrate. In the composting process, under aerobic conditions, thermophilic biodegradation of organic wastes at $40 \pm 60\%$ moisture content occurs to form relatively stable, humus-like material [22]. The process is expressed as:

Organic matter $+ O_2 + \text{microorganisms} \rightarrow CO_2 + H_2O + H_2O$

The process steps followed during active composting are follows:

- (1) Windrow making.
- (2) Drying to reduce moisture content to 5 + 60%
- (3) Trimming for breaking of lumps.
- (4) Aerotilling to loosen the presspera substrate
- (5) Inoculating with microbial count of for faster dependent of the statistic of the statis
- (6) Aerotilling for proper dxing, distribution and oxygenation.
- (7) Maintaining incomum by a faction of small doses at regular intervals for better or fernance.
- (8) Efficient comptrates pentwas' praying.
- (9) Temperate multiined typen 50 and 70° C.
- (10) Moi are content maintaine. In the range of $40 \pm 60\%$.
- (11) Aeration content of the each concentrated spentwash approximiting \rightarrow aerotilling to attain the oxygen concentration more than 10%.
- (12) Climatic conditions including temperature, wind velocity and rainfall taken into the consideration.
- (13) Improper composting encourages breeding of flies, which became a nuisance. Fly control achieved through grinding, turning and heating effect in proper decomposing conditions.
- (14) Nitrogen loss in the form of ammonia in anaerobic conditions prevented by maintaining proper initial C:N ratio (range 38 ± 40), pH, moisture content and oxygen concentration.

3.1.1.2. Activated sludge process. The most common postbiomethanation step is the activated sludge process wherein research efforts are targeted at improvements in the reactor configuration and performance. For instance, aerobic sequencing batch reactor (SBR) is reported to be a promising solution for the treatment of effluents originating from small wineries [23]. The treatment system consisted of a primary settling tank, an intermediate retention trough, two storage tanks and an aerobic treatment tank. A start up period of 7 days was given to the aerobic reactor and the system resulted in 92% COD and 97.5% BOD removal.

The activated sludge process and is variations lize mixed cultures. To enhance the efficiency of probics systems, several workers have focused on the catment v pure c ures. Further, aerobic treatment has also been exall edge a precursor to anaerobic treatment studies in both be spentwash and molasses, aerobic protrease profibeet spottwash with *Penicil*sulted in put 74% eduction in phenolics lium decumbens [4]. Anaerobic digestion content and 4 duction in or natment restricted in a sharp drop in COD without aer of pre removal efficiencies w decreasing hydraulic retention time (HR^T rm organic matter removal was marginally higher for molasses previously fermented with P. decumbens. The be robic reaction followed first-order kinetics and the rate conan decreased or hereasing the organic loading with untreated sta hower r, it remained almost constant with pre-treated mola. polasses 2,25]. *Geotrichum candidum* is another species that in partial elimination of phenolic inhibitors such as genre sic acto, gallic acid, quercetin, p-coumaric acid, etc., thereby enhancing the effectiveness of anaerobic process [26].

3.1.1.3. Constructed wetlands (CWs). Billore et al. [27] have demonstrated a four-celled horizontal subsurface flow (HSF) CW for the treatment of distillery effluent after anaerobic treatment. The post-anaerobic treated effluent had a BOD of about 2500 mg/l and a COD of nearly14,000 mg/l. A pre-treatment chamber filled with gravel was used to capture the suspended solids. All the cells were filled with gravel up to varying heights and cells three and four supported the plants Typha latipholia and Phragmites karka, respectively. The overall retention time was 14.4 days and the treatment resulted in 64% COD, 85% BOD, 42% total solids and 79% phosphorus content reduction. In another study, a distillery in northern India is presently employing CW for polishing the effluent prior to land discharge for irrigation in the surrounding paddy fields. The effluent was initially subjected to primary treatment which included settling and anaerobic digestion in a structured media attached growth (SMAG)-type anaerobic reactor. The primary treated effluent, with a COD of 28,000-35,000 mg/l, is subjected to two-stage aeration to bring down the COD to 400 mg/l. Thereafter, it is directed to a CW before final discharge.

3.1.2. Anaerobic process

The high organic content of molasses spentwash makes anaerobic treatment attractive in comparison to direct aerobic treatment. Therefore, biomethanation is the primary treatment step and is often followed by two-stage aerobic treatment before

Table 4					
Performance of various anaerobic fluidized-bed reactor for treating molasses distillery wastewater					
Name of the Reactor	OLR (Kg COD/m ³ /day)	HRT (days)	COD removal (%)		
Down-flow anaerobic fluidized-bed reactor	3.27–5.75	1.35–O.87	70-81		
Inverse turbulent bed reactor	15		75 - 85		
Inverse fluidized-bed reactor	10-12	60–3	>90		
Anaerobic fixed film reactor	10		60–70		
Granular-bed anaerobic bed reactor	4.75		80		
Anaerobic contact filer	14	4	73–97		
Fluidized-bed reactor	2.55-0.37	5.88	96.5		
Inverse anaerobic fluidized-bed reactor	34.05	0.19	84		

Per

discharge into a water body or on land for irrigation [9]. Aerobic treatment alone is not feasible due to the high-energy consumption for aeration, cooling, etc. Moreover, 50% of the COD is converted to sludge after aerobic treatment [28]. In contrast, anaerobic treatment converts over half of the effluent COD into biogas [6]. Anaerobic treatment could be successfully operated at high organic loading rates; also, the biogas thus generated can be utilized for steam generation in the boilers, thereby meeting the energy demands of the unit [9].

Inv

Inv

Fh

The performance and treatment efficiency of anaerobic process could be influenced both by inoculums source and feed pre-treatment. In particular, thermal treatment of wastewaters could result in rapid degradation of organic matter leading to lower HRT, higher loading rate and BOD reduction. Moreover, the methane content and calorific value of biogas produced thermophilic systems were higher [29]. Significant impl ement was observed using inoculums from anaerobic lagoon w h 27.2% COD reduction with thermally pre-tra stewat and 51% reduction with untreated wastewater Furthe the tem idered as perature of thermal pre-treatment was also to be co an important one. After 150 days additation od, wastewn ater treated at lower temperature 1° C) should 66% COD reduction [30]. This was nearly that he removal a pined with micronutrients (iron, treatment at 23 °C. Further, addition boron and molybdenum) minated the g adaptation periods. Anaerobic lagoon are the simplest option for the anaerobic spentway. Though anaerobic lagoons are treatment of distiller dist teries, high rate anaerobic reactors still employed in Ind. are more popul [31]. T. e reactor offer the advantage of sepn solid, set in don time (SRT) so that slow arating the **r**RT fi growing anaerobi microorga is sms can remain in the reactor indepen nt of low.

d reactor. This involves immobilization of 3.1.2.1. Fixea some inert support to limit the loss of microorganisms biomass and enhance the bacterial activity per unit of reactor volume. Moreover, it provides higher COD removal at low HRT and better tolerance to toxic and organic shock loadings. In anaerobic contact filters, various packing materials, viz. polyurethane, clay brick, granular activated carbon (GAC), polyvinyl chloride (PVC) plastic media have been employed resulting in 67-98% reduction in COD [32-35]. GAC as support media is relatively expensive, but because of its adsorptive properties, it contributes towards improved process stability. The interference by sulfate, unionized sulfate and total hydrogen sulfide is anaerobic.

It was observed that the percenta, sulfate rem al increased with increasing HRT from 2.0 5 day. This might be due to the utilization of sulfate as mutrient by heroor anisms present in anaerobic contact and alter teir converton to sulfide by (S 3) under anaerobic conditions. sulfate reducing hacte r sulfate acentration (426 mg/l), the removal However, at high bly due to low Repopulation in comparison to decreased, p methanog As. Ak the removal sulfide was explained by stripping of hydrogen stande from liquid to vapor phase by carbon dia d methane generated during the anaerobic process. In other study [36], anaerobic treatment of undiluted whisky pot e using an upper ow anaerobic filter (UAF) packed with special port type realited in 76% COD removal. The pilot system sonsister of a decanter, dephosphatation or magnesium phosphate (MAP) (MgNH₄PO₄) reactor, denitrifiammon n reactor, nitrification reactor and sedimentation tank for the reduction of nitrogen and phosphate. Down-flow filter using plastic PVC as support material has been employed for the treatment of beet molasses wastewater [37]. The system resulted in 55-85% reduction in COD. Also, though high sulfide concentration (4250 mg) was inhibitory to the system, it was not toxic at higher loadings (44 kg COD/m³/day) probably due to high stripping of H₂S.

3.1.2.2. Anaerobic fluidized-bed reactor. An anaerobic fluidized-bed reactor (AFBR) consists of a vertical vessel containing an inorganic media (e.g. rock, sand, activated carbon, anion and cation exchange resins) [38]. Anaerobic fluidized-bed reactor utilizes small fluidized media particles to induce extensive cell immobilization thereby achieving a high reactor biomass hold-up and a long mean cell residence time [39]. The fluidized-bed technology presented a series of advantages compared to other kinds of anaerobic processes [40], like high organic loading rates and short hydraulic retention times. Therefore, a number of design modifications have been tested or adopted in order to improve the performance of the system. In the classic case of fluidized systems, the solid particles have had a higher density than the fluid. Performance of various anaerobic fluidized-bed reactors for treating molasses distillery wastewater is given in Table 4.

3.1.2.3. Inverse anaerobic fluidized-bed reactor. In the field of anaerobic treatment process, an inverse anaerobic fluidized-bed reactor (IAFBR) has emerged as a good alternative for the treatment of spentwash from distilleries. In an inverse fluidization,

References

[41]

[43] [44]

[45]

[46]

[35]

[47]

[7]

the liquid specific density was found to be higher than the specific density of the particle, and the bed has been expanded downward by the liquid flow. The down-flow fluidized-bed reactor or inverse fluidized-bed reactor has been described for application in the anaerobic treatment of wastewater [41]. In their description, of down-flow fluidization, particles with a specific gravity smaller than the liquid are fluidized downward by a concurrent flow of liquid. The biofilm formation and its effect on hydrodynamics of the reverse fluidized-bed reactor have been described [41]. The application of inverse fluidization in wastewater treatment from laboratory to full-scale bioreactors has been described [42]. The biofilm, growing on the surface of support particles, increased the overall bioparticle (support particle plus biofilm) diameter. It resulted in bed expansion and very slow movement of the lower bed level downward until the lower bed level reached the lower draft tube opening and some of the bioparticles entered the draft tube with the liquid flow. The inverse fluidized-bed biofilm reactor designed in such a way that the biofilm thickness can be controlled to avoid the intrabiofilm diffusion limitations.

Perlite was an interesting carrier material when compared to others like cork, polyethylene or polypropylene [41]. The carrier particles chosen in a particular study [7] were perlite. Before starting up the reactor, physical properties of the carrier material were determined. 1 mm diameter perlite particle was found to have a wet specific density of 295 kg/m^3 with specific surface area of 7.010 m²/g and low energy requirements for fluidization It was used for the treatment of distillery waste and performastudies were carried out for 65 days. Once the down-flow anae obic fluidized-bed system reached the steady state, the organi load was increased stepwise by reducing HRT 0.19 days, while maintaining the constant feed of OD cortion. Most particles have been covered with thin b entra-NE uniform thickness. This system achieve 4%emoval at 7]. an organic loading rate (OLR) of 35^{1} COD/m³/s.

3.2. Physico-chemical treatment

entwash after biological treatment Sugarcane molasses by both anaerobic are aerobic ethod can still have a BOD o, en though biological treatment of 250–500 mg/l [11]. COD noval, the effluent still retains the results in signif J. The color hearing melanoidins are barely dark color conventinal biologn at treatment such as methane affected b fermentati and sludge process [49]. Further, multistage bi fical treatment reduces the organic load but intensifies the conductor due to re-polymerization of colored compounds [50]. In this chtext, various physico-chemical treatment options have been explored.

3.2.1. Adsorption

Activated carbon is a widely used adsorbent for the removal of organic pollutants from wastewater but the relatively high cost restricts its usage. Decolorization of synthetic melanoidin using commercially available activated carbon as well as activated carbon produced from sugarcane bagasse was investigated by [51]. The adsorptive capacity of the different activated carbons was found to be quite comparable. Chemically modified

bagasse using 2-diethylaminoethyl-chloride hydrochloride and 3-chloro-2-hydroxypropyltrimethylammonium chloride was capable of decolorizing diluted spentwash [52]. Significant decolorization was observed in packed bed studies on anaerobically treated spentwash using commercial activated charcoal with a surface area of 1400 m²/g [53]. Almost complete decolorization (>99%) was obtained with 70% of the diluted sample, which also displayed over 90% BOD and COD removal. In contrast, other workers have reported adsorption by activated carbon to be ineffective in the treatment of distillanceffluent [54,55]. Adsorption by commercially available owdere ctivated carbons resulted in only 18% color regoval, however, combined treatment using coagulation-floccul on with pollectrolyte followed by adsorption result in almost complete ecolorization [54,56] reported color emoval up to 1% th pyorchar. Yet another adsorbent, at has but examined is the natural carbohydrate polymer che san derived from the exoskeleton of crustaceans [51, studied aring the reatment of distillery wastewater using hitosan as a prior exchanger. At an optimum dosage of 10, and 30 min contact time, 98% color and 99% COD removal was bserved.

3.2 . Coagulation and flocculation

nanc et al. [4] reported that coagulation with alum and alts was no effective for color removal. They explored irol ozone eatment with anaerobically digested effluent. lime The optimum dosage of lime was found to be 10 g/l result-2.5% COD removal and 67.6% reduction in color in 11 30 min period. These findings have been in disagreement with those of [49] who used a commercial inorganic flocculent, a polymer of ferric hydroxysulfate with a chemical formula $[Fe_2(OH)n \cdot (SO4)_{3-n/2}]$ m for the treatment of molasses wastewater. The treatment resulted in around 87% decolorization for biodigested effluents; however an excess of flocculent hindered the process due to an increase in turbidity and TOC content. FeCl₃ and AlCl₃ were also tested for decolorization of biodigested effluent and showed similar removal efficiencies. About 93% reduction in color and 76% reduction in Total Organic Carbon (TOC) were achieved when either FeCl₃ or AlCl₃ was used alone. The process was independent of chloride and sulfate ion concentration but was adversely affected by high fluoride concentration. However, in the presence of high flocculent concentration (40 g/l), addition of 30 g/l CaO enhanced the decolorization process resulting in 93% color removal. This was attributed to the ability of calcium ions to destabilize the negatively charged melanoidins; further, formation of calcium fluoride (CaF2) also precipitated the fluoride ions. Almost complete color removal (98%) of biologically treated distillery effluent has been reported with conventional coagulants such as ferrous sulfate, ferric sulfate and alum under alkaline conditions [58]. The best results were obtained using Percol 47, a commercial organic anionic polyelectrolyte, in combination with ferrous sulfate and lime. The combination resulted in 99% reduction in color and 87 and 92% reduction in COD and BOD, respectively. Similar findings have also been reported by [55]. Coagulation studies on spentwash after anaerobic-aerobic treatment have also been conducted using bleaching powder followed

by aluminum sulfate [59]. The optimum dosage was 5 g/l bleaching powder followed by 3 g/l of aluminum sulfate that resulted in 96% removal in color, accompanied by up to 97% reduction in BOD and COD. Non-conventional coagulants namely wastewater from an iron pickling industry which is rich in iron and chloride ions and titanium ore processing industry containing significant amounts of iron and sulfate ions have also been examined [58]. The iron pickling wastewater gave better results with 92% COD removal, combined with over 98% color removal. Though the titanium processing wastewater exhibited similar color removal levels, the COD and BOD reductions were perceptibly lower.

3.2.3. Oxidation process

Ozone destroys hazardous organic contaminants and that have been applied for the treatment of dyes, phenolics, pesticides, etc. [50]. Oxidation by ozone could achieve 80% decolorization for biologically treated spentwash with simultaneous 15-25% COD reduction. It also resulted in improved biodegradability of the effluent. However, ozone only transforms the chromophore groups but does not degrade the dark colored polymeric compounds in the effluent [60,50]. Similarly, oxidation of the effluent with chlorine resulted in >97% color removal but the color reappeared after a few days [55]. Ozone in combination with UV radiation enhanced spentwash degradation in terms of COD; however, ozone with hydrogen peroxide showed only marginal reduction even on a very dilute effluent [61]. Sal by exposed to 2 h ultrasound pre-treatment displayed 44% D removal after 72 h of aerobic oxidation compared to 25% C D reduction shown by untreated samples. These SUL are col trary to those of [55] who concluded ultrast ac treat ineffective for distillery spentwash treats at. A nt to be ion of wet air oxidation and adsorption been essfully used to demonstrate the removal of sulfies from divery wastewater. Studies were done in a contract urrent reaction ontaining 25 cm base of small crushed stones superting a 20 cm column of bagasse ash as an adsort int [62]. The wave ewater was applied from the top of the region and air was supplied at the rate of 1.0 l/min. The treat and remarked 57% COD, 72% BOD, 83% Vet a oxidatic has been recommended TOC and 94% sulfates ed picks scher for treating anaerobically as part of a co [63]. The organaerobic effluent was therdigested s ntwas -treated $150 \,^{\circ}\text{C}$ uncer pressure in the absence of air. mally p This was lloy me treatment, after which the efflua 2 h wet oxidation at 225 °C. 95% color removal ent underw his scheme. Another option is photo catalytic was obtained oxidation that have been studied using solar radiation and TiO₂ as the photo catalyst [64]. Use of TiO_2 was found to be very effective as the destructive oxidation process leads to complete mineralization of effluent to CO_2 and H_2O . Up to 97% degradation of organic contaminants was achieved in 90 min [65] studied combined electron beam and coagulation treatment of distillery slops from distilleries processing grain, potato, beet and some other plant materials. Humic compounds and lignin derivatives constitute the major portion of this dark brown wastewater. The distillery wastewater was diluted with municipal wastewater in the ratio of 3:4, irradiated with electron beam and then coagulated with $Fe_2(SO_4)_3$. The optical absorption in UV region was decreased by 65–70% after this treatment.

3.2.4. Membrane treatment

Pre-treatment of spentwash with ceramic membranes prior to anaerobic digestion is reported to halve the COD from 36,000 to 18,000 mg/l [66]. The total membrane area was 0.2 m^2 and the system was operated at a fluid velocity of 6.08 m/s and 0.5 bar transmembrane pressure. In addition to COD reduction, the pre-treatment also improved the efficiency of the anaerobic process possibly due to the removed of inducting substances [67] employed emulsion liquid methorane (ELM echnique in a batch process for spentwash treatment. Water-oil vater type of emulsion was used to separate and contraste the olutes resulting in 86% and 97% declease in COD. B. B. D, respectively. Electro dialysis has be dexplore for desals of spentwash using cation and anion exchange cembrane resulting in 50–60% assium content [69] In another study [69], reduction in p ment of vina when the molasses by electro reported the dialysis using a su less steel canode, titanium alloy anode and 4% w/v NaCl as electrolytic agent. Up to 88% COD reduction 19.5 was obtained, owever, the COD removal percentage at creased at higher wastewater feeding rates. In addition, reverse mosis (RO) has also been employed for distillery wastewater tment. A up in western India is currently processing efflutained der anaerobic digestion, followed by hold-up in en a tank munitained under aerobic conditions, in a RO system Agrawal, personal communication). 290 m³/day of RO treated effluent is mixed with 300 m³/day of fresh water and used in the process for various operations like molasses dilution $(290 \text{ m}^3/\text{day})$, make-up water for cooling tower $(178 \text{ m}^3/\text{day})$, fermenter washing (45 m³/day), etc. Yet another unit in southern India is employing disc and tube RO modules for direct treatment of the anaerobically digested spentwash (M. Prabhakar Rao, personal communication). The permeate is discharged while the concentrate is used for biocomposting with sugarcane press mud. In a recent study [70], reported pilot trials on distillery spentwash using a hybrid nano filtration (NF) and RO process. Both the NF and RO stages employed thin film composite (TFC) membranes in spiral wound configuration with module dimensions of 2.5 in. diameter and 21 in. length. NF was primarily effective in removing the color and colloidal particles accompanied by 80, 95 and 45% reduction in total dissolved solids (TDS), conductivity and chloride concentration, respectively, at an optimum feed pressure of 30-50 bar. The subsequent RO operation at a feed pressure of 50 bar resulted in 99% reduction each in COD, potassium and residual TDS.

3.2.5. Evaporation/combustion

Molasses spentwash containing 4% solids can be concentrated to a maximum of 40% solids in a quintuple-effect evaporation system with thermal vapor recompression [71,72]. The condensate with a COD of 280 mg/l can be used in fermenters. The concentrated mother liquor is spray dried using hot air at 180 °C to obtain a desiccated powder with a calorific value of around 3200 kcal/kg. The powder is typically mixed with 20% agricultural waste and burnt in a boiler. The use of recirculating fluidized-bed (RCFB) incinerator is recommended to overcome the constraints due to stickiness of spentwash and its high sulfate content [73]. Combustion is also an effective method of on-site vinasse disposal as it is accompanied by production of potassium-rich ash [19] that can be used for land application.

4. Discussion

The various biological and physico-chemical treatment methods have been identified for the treatment of wastewater from molasses-based distilleries. Due to the high COD of raw spentwash, application of anaerobic treatment technology (involving biogas recovery) has been reported to be highly effective which is considered as the first step. [74–77,33]. The treatment with anaerobic lagoon and conventional digesters are applied to some extent in the field. However, most distilleries in India employ high rate digesters wherein the HRT is decoupled from the biological/solids retention time (BRT/SRT) thereby retaining the slow growing anaerobic microbes in the reactor even at high wastewater flow.

Biological treatment using aerobic processes like activated sludge, biocomposting, etc. is presently practised by various molasses-based distilleries. Due to the large volumes generated, only a part of the total spentwash gets consumed in biocomposting. Biocomposting utilizes sugarcane pressmud as the filler material; thus it is typically employed by distilleries attached to sugar mills. Since sugar manufacturing is a seasonal operation, pressmud availability is often a constraint. Further, biocompos ing requires a large amount of land; also, it cannot be carried ou during the rainy season. Though aerobic treatmer лКС e conventional activated sludge process leads to sign dcant re in COD, the process is energy intensive and a color r lction still inadequate. Anaerobic reactor, partice arly, rse anaerobic fluidized-bed reactor [7] appeared be the one choice for the reduction in pollutional strep a we distilleries we n compared to other reactors. However, proceedesign and control criteria for full exploitation these reactors vet to be developed.

In general physic chemic treatment has met with little success. The tecking such as adsorption, coaguladation, ocesses and membrane treatment tion/flocculation are more effective in emoval for or as well as reduction in organic le ring, slue e production and disposal is a constraint in coagula. v/flo alan adsorption. Settling is found to even with the addition of coagulants and other be unsatisfact lum, lime, ferric chloride, other aluminum additives such as salts, iron salts, etc in the coagulation/flocculation process. Also, the cost of chemicals, adsorbents and membranes is deterrent to the adoption of these methods [78]. Membrane operations like microfiltration/ultrafiltration for spentwash treatment are characterized by significant membrane fouling that limits its applicability [79].

5. Conclusion

This review article examined the extent of pollution created by distilleries and the different methods available for the treatment and disposal of distillery wastewater. Physico-chemical methods are capable of reducing organic load, consequently, in spite of cost; the advanced methods like membrane filtration, oxidation by ozone are being field trials. Biological treatment methods appear to be only choice for the removal of color and organic content; however, some of the questions are yet to be answered on its process efficiency. This is because of the lack of information on various aspects such as desirable influent COD, optimal level of volatile fatty acids (VFA) concentration in the reactor, the reliable estimates of the bio kinetics constants and their dependence on the substrate level (COD or VFA).

This review has showed, the actor namely in inverse anaerobic fluidized-bed technology been found be more effective than anaerobic filter tehnology is it favore, the transport of microbial cells from the bulk to the synace and thus enhanced the contact between mice organisms and the substrate for the treatment of high vertice waster from molassesbased distilleries one review the indicated the real evidence about exceller $r_{\rm P}$ vsical characteristics and adequate environment of perce as a maerobic moroorganism immobilizer in an inverse anaerobic field is bed reactor and presented series of advantages of an inversion naerobic fluidized-bed technology pared to other kinds of anaerobic processes interms of high co nic loading thes and short hydraulic retention times. The or approximation of this reactor to the treatment of molasses-based waster der would give satisfactory results compared distin o other brongical reactors.

References

- K.A. Subramanian, S.K. Singal, M. Saxena, S. Singal, Utilization of liquid biofuels in automotive deselengines: an Indian perspective, Biomass Bioenergy 29 (1) (2005) 65–72.
- [2] S. Mahimairaja, N.S. Bolan, Problems and prospects of agricultural use of distillery spentwash in India, in: Third Australian and New Zealand Soil Science Societies Joint Conference, 5–9 December, Australia, 2004.
- [3] F.J. Fitzgibbon, D. Singh, G. Mcmullan, R. Marchant, The effect of phenolic acids and molasses spentwash concentrations on distillery wastewater remediation by fungi, Process Biochem. 33 (8) (1998) 799–803.
- [4] J. Uppal, Water utilization and effluent treatment in Indian alcohol industry: an overview, in: P.K. Tewari (Ed.), Liquid Asset. Proceeding of Indo-EU Workshop on Promoting Efficient Water Use in Agro-Based Industries, TERI Press, New Delhi, India, 2004, pp. 13–19.
- [5] G.J. Sheehan, P.F. Greenfield, Utilization, treatment and disposal of distillery wastewater, Water Res. 14 (3) (1980) 257–277.
- [6] A.C. Wilkie, K.J. Riedesel, J.M. Owens, Stillage characterization and anaerobic treatment of ethanol stillage from conventional and cellulosic feedstocks, Biomass Bioenergy 19 (2) (2000) 63–102.
- [7] R. Sowmeyan, G. Swaminathan, Inverse anaerobic fluidized bed reactor for treating high strength organic wastewater, Bioresour. Technol. 99 (2008) 3877–3880.
- [8] B.G. Yeoh, Two-phase anaerobic treatment of cane-molasses alcohol stillage, Water Sci. Technol. 36 (6–7) (1997) 441–448.
- [9] T. Nandy, S. Shastry, S. Kaul, Wastewater management in cane molasses distillery involving bioresource recovery, J. Environ. Manage. 65 (1) (2002) 25–38.
- [10] P.U. Patil, B.P. Kapadnis, V.S. Dhamankar, Decolorization of synthetic melanoidin and biogas effluent by immobilized fungal isolate of *aspergillus niger* UM2, All India Distillers's Association (AIDA) (2003), Newsletter (2003) 53–56.
- [11] I.D. Mall, V. Kumar, Removal of organic mater from distillery effluents using low cost adsorbent, Chem. Eng. World XXXII (7) (1997) 89–96.

- [12] N. Jain, A.K. Minocha, C.L. Verma, Degradation of predigested distillery effluent by isolated bacteria strains, Indian J. Exp. Biol. 40 (2002) 101–105.
- [13] D.F. Kalavathi, L. Uma, G. Subramanian, Degradation and metabolization of the pigment-melanodin in a distillery effluent by the marine cyanobactrium oscillatoria boryana BDU 92181, Enzyme Microb. Technol. 29 (4–5) (2001) 246–251.
- [14] M.D. Rivero-Pei rez, M. Pei rez, M.L. Jose, Role of melanoidins in sweet wines, Anal. Chim. Acta 458 (1) (2002) 169–175.
- [15] T. Gonzalez, M.C. Terron, S. Yague, E. Zapico, G.C. Galletti, A.E. Gonzalez, Pyrolysis/gas chromatography/mass spectrometry monitoring of fungal biotreated distillery wastewater using *trametes* sp.I-62 (CECT 20197), Rapid Commun. Mass Spectrom. 14 (15) (2000) 1417–1424.
- [16] S. Sirianuntapiboon, P. Zohsalam, S. Ohmomo, Decolorization of molasses wastewater by citeromyces sp WR-43-6, Process Biochem. 39 (8) (2004) 917–924.
- [17] M.A. Godshall, Removal of colorants and polysaccharides and the quality of white sugar, in: Proceedings of Sixth International Symposium Organized by Association Andrew van Hook (avh), March 25, Reims, France, 1999, pp. 28–35.
- [18] G.R. Pathade, in: P.K. Goel (Ed.), A review of current technologies for distillery wastewater treatment, Advances in Wastewater Treatment Techno Science Publication, Rajastan, India, 1999, pp. 180–239.
- [19] L.A.B. Cortez, L.E.B. Pereiz, Experiences in vinasse disposal: Part III: combustion to vinasse # fuel oil emulsion, Braz. J. Chem. Eng. (1997), http://www.scielo.br/scielo.php.
- [20] J.G.O. Rodrýĭguez, Effects of vinasse on sugarcane (saccharum officinarum) productivity, revista de la facutlad de agronomia, universidad del zulia 17 (2000) 1710–1716.
- [21] H.C. Joshi, Bio-energy potential of distillery effluents, Bio Energy News 3 (3) (1999) 10–15.
- [22] P.N. Cheremisinoff, Biomanagement of wastewater and waster, PTR-Prentice Hall Inc., New Jersey, 1994, pp. 162–183.
- [23] M. Torrijos, R. Moletta, Winery wastewater depollution by sequencing batch reactor, Water Sci. Technol. 35 (1) (1997) 249–257.
- [24] A.M. Jimeĭnez, R. Borja, A. Martin, Aerobic–anaerobiol Tegradation beet molasses alcoholic fermentation wastewater, Proc. 88 Bit mem. 38 ((2003) 1275–1284.
- [25] A.M. Jimešnez, R. Borja, A. Martin, A comparative kinetic reduction of the anaerobic digestion of untreated molastic and collastic previously. mented with pencillium decumbens in here reactor, whether. Eng. J. 18 (2) (2004) 121–132.
- [26] R. Borja, A. Martin, R. Maestrout, Doub, M.M. Durah, chancement of the anaerobic digestion of wine distillery estewater by the removal of phenolic inhibitors, Biores a. Technol. 45 (2), 1933 99–104.
- [27] S.K. Billore, N. Singh, J.K. Ram, J.K. Sharma, Singh, R.M. Nelson, P. Doss, Treatment of dolasses based distillery effluent in a constructed wetland in central New, Water L. Technol. 44 (11–12) (2001) 441–448.
- [28] T. Sennitt, Emissions Theory omics of big has and power, in: 68th Annual Water Industry Engineer, and Operator Conference, Schweppes Centre, Bendigory & September, 20
- [29] A.Vli Caesa.I. Zerboulis, The manilic anaerobic digestion of alcohol discury wastever Pioresour. Jechnol. 43 (2) (1993) 131–140.
- [30] G.M. Ver, J.F. ampn, and and K.S.S. Lele, J.B. Joshi, Overall treatment of therma pre-treated distillery waste Part II, Indian Chem. Eng. Sect A 40 (3) (1998) 32–240.
- [31] K. Lata, Kan, M. Balakrishnan, K.V. Rajeshwari, V.V.N. Kishore, Assessment of Bromethanation Potential of Selected Industrial Organic Effluents in India, Resour. Conserv. Recycl. 35 (3) (2002) 147–161.
- [32] A. Bories, J. Raynal, F. Bazile, Anaerobic digestion of anaerobic of high strength distillery wastewater (cane molasses stillage) in a fixed-film reactor, Biological Wastes 23 (4) (1998) 251–267.
- [33] R. Seth, S.K. Goyal, B.K. Handa, Fixed film biomethanation of distillery spentwash using low cost porous media, Resour. Conserv. Recycl. 14 (1995) 79–85.
- [34] S.K. Goyal, R. Seth, B.K. Handa, Diphasic fixed-film biomethanation of distillery spentwash, Bioresour. Technol. 56 (2–3) (1996) 239–244.
- [35] K. Vijayaraghavan, T.K. Ramanujam, Performance of anaerobic contact filter in series for treating distillery spentwash, Bioprocess Biosyst Eng 22 (2) (2000) 109–114.

- [36] M. Tokuda, Y. Fujiwara, K. Kida, Pilot plant test for the removal of organic matter, n and p from whisky potable, Process Biochem. 35 (3–4) (1999) 267–275.
- [37] N. Athanasopoulos, Anaerobic treatment of beet molasses alcoholic fermentation wastewater in a downflow filter, Resour. Conserv. 15 (1–2) (1987) 147–150.
- [38] Metcalf and Eddy, Inc., Wastewater Engineering: Treatment, Disposal and Reuse, Tata Mcgraw-Hill Publishing Company Limited, New Delhi, 2003.
- [39] W.K. Shieh, Y. Hsu, Biomass loss from an anaerobic fluidized bed reactor, Water Res. 30 (1996) 1253–1257.
- [40] V. Diez-Blanco, P. Garcia-Encina, F. Fernandez-Polanco, Effect of biofilm growth, gas, liquid velocities on the expansion of an anaerobic fluidized bed reactor, Water Res. 29 (1995) 1649 (2014)
- [41] D. Garcia-Calderon, P. Buffiere, B. Joletta, S. Elharch, Influence of biomass accumulation on bed expansion characteristic of a down-flow anaerobic fluidized-bed reactor Biotech. J. Bioeng. 57, 198) 136–144.
- [42] D. Karamanev, G. Nikolov averse fluidizion in we ewater treatment, Environ. Prog. 15 (1996) 24–196.
- [43] P. Buffiere, Jean-Pierer Bergeon, J. Moletta, The inverse turbulent bed: a novel bioreactor for an object atment, Water Res. 34 (2000) 673–677.
- [44] C. Arnaiz, S. Johaleh, J. Johato, R. Monda, Start up of an anaerobic inverse turbe int bed reactor of with one distillery wastewater using pre-color ed particles, Water pre-chnol. 51 (2005) 153–158.
- [45] A. Gan, agni Rao, Venkata Naidu, K. Krishna Prasad, N. Chandrasekhar Rao, S. Venkata Mon, Annapurna Jetty, P.N. Sarma, Anaerobic treatment was ewater with high spended solids from a bulk drug industry using fixed film reactor (AFFR), Bioresour. Technol. 96 (1) (2005) 87–93.
- J.C. Akunnagi I. Clark, Performance of a granular-bed anaerobic baffled reactor (grabb treating whisky distillery wastewater, Bioresour. Technol. 74 (3) (2000) 17–261.
- [47] Perez, J. Romero, D. Sales, Anaerobic thermophilic fluidized bed treas. So of industrial wastewater: effect of F:M relationship, Chemosphere 38 (14) (1999) 3443–3461.
- [48] Anne, F. Ciner, I. Ozturk, Colour removal from fermentation industry effluents, Water Sci. Technol. 40 (1) (1999) 331–338.
- [49] V.P. Migo, M. Matsumara, E.J.D. Rosaria, H. Kataoka, Decolorization of molasses wastewater using an inorganic flocculant, J. Fermentation Bioeng. 75 (6) (1993) 438–442.
- [50] M. Peña, M. Coca, R. Gonzalez, R. Rioja, M.T. Garcia, Chemical oxidation of wastewater from molasses fermentation with ozone, Chemosphere 51 (9) (2003) 893–900.
- [51] E.C. Bernardo, R. Egashira, J. Kawasaki, Decolorization of molasses wastewater using activated carbon prepared from cane bagasse, Carbon 35 (9) (1997) 1217–1221.
- [52] J.D. Mane, S. Modi, S. Nagawade, S. Nagawade, S.P. Phadnis, V.M. Bhandari, Treatment of spentwash ushing chemically modified bagasse and colour removal studies, Bioresour. Technol. 97 (14) (2006) 1752–1755.
- [53] R. Chandra, P.K. Pandey, Decolorization of anaerobically treated distillery effluent by activated charcoal adsorption method, Indian J. Environ. Prot. 21 (2) (2000) 134–137.
- [54] D. Sekar, D.V.S. Murthy, Colour removal of distillery spentwash by adsorption technique, Indian Chem. Eng. Sect. A 40 (4) (1998) 176–181.
- [55] A. Mandal, K. Oiha, D.N. Ghosh, Removal of colour from distillery wastewater by different processes, Indian Chem. Eng. Sect. B 45 (4) (2003) 264–267.
- [56] D.S. Ramteke, S.R. Wate, C.A. Moghe, Comparative adsorption studies of distillery waste on activated carbon, Indian J. Environ. Health 31 (1) (1989) 17–24.
- [57] I.G. Lalov, I.I. Guerginov, A. Krysteva, K. Fartsov, Treatment of wastewater from distilleries with chitosan, Water Res. 34 (5) (2000) 1503–1506.
- [58] R.A. Pandey, S. Malhotra, A. Tankhiwale, S. Pande, P.P. Pathe, S.N. Kaul, Treatment of biologically treated distillery effluent—a case study, Int. J. Environ. Stud. 60 (3) (2003) 263–275.
- [59] R. Chandra, H. Singh, Chemical decolorization of anaerobically treated distillery effluent, Indian J. Environ. Prot. 19 (11) (1999) 833–837.
- [60] C.G. Alfafara, V.P. Migo, J.A. Amrante, R.F. Dallo, M. Matsumara, Ozone treatment of distillery slop waste, Water Sci. Technol. 42 (3–4) (2000) 193–198.

- [61] F.J. Beltran, J.M. Encinar, J.F. Gonzalez, Industrial wastewater advanced oxidation. Part 2. ozone combination with hydrogen peroxide or UV radiation, Water Res. 31 (10) (1997) 2415–2428.
- [62] R.W. Gaikwad, P.K. Naik, Technology for the removal of sulfate from distillery wastewater, Indian J. Environ. Prot. 20 (2) (2000) 106–108.
- [63] A.D. Dhale, V.V. Mahajani, Treatment of distillery waste after bio-gas generation: wet oxidation, Indian J. Chem. Technol. 7 (2000) 11–18.
- [64] A.K. Kulkarni, Solar assisted photocatalytic oxidation of distillery waste, Indian Chem. Eng. 40 (2) (1998) 169–172.
- [65] A.K. Pikaev, A.V. Pomomarev, A.V. Bludenko, V.N. Minin, L.M. Elizar'eva, Combined electronic-beam and coagulation purification of molasses distillery slops. Features of the method, technical and economical evaluation of large scale facility, Rad. Phys. Chem. 61 (1) (2001) 81–87.
- [66] I.S. Chang, K.H. Choo, C.H. Lee, U.H. Pek, U.C. Koh, S.W. Kim, J.H. Koh, Application of ceramic membrabe as a pre-treatment in anaerobic digestion of alcohol-distillery wastes, J. Membrane Sci. 90 (1–2) (1994) 131–139.
- [67] T. Kumaresan, K.M.M. Sheriffa Begum, P. Sivashanmugam, N. Anantharaman, S. Sundram, Experimental studies on treatment of distillery effluent by liquid membrane extraction, Chem. Eng. J. 95 (1–3) (2003) 99–204.
- [68] F.G.N. De Wilde, Deminerlization of a molasses distillery wastewater, Desalination 67 (1987) 481–493.
- [69] A.G. Vlyssides, C.J. Israilides, M. Loizidou, G. Karvouni, V. Maurafeti, Electrochemical treatment of vinasse from beet molasses, Water Sci. Technol. 36 (2–3) (1997) 271–278.
- [70] S.K. Nataraj, K.M. Hosamani, T.M. Aminabhavi, Distillery wastewater treatment by the membrane-based nano-filtration and reverse osmosis processes, Water Res. 40 (12) (2006) 2349–2356.

- [71] H.C. Bhandari, A.K. Mitra, S. Kumar, Crest's integrated system: reduction and recycling of effluents in distilleries, in: P.K. Tewari (Ed.), Liquid Asset. Proceeding of Indo-EU Workshop on Promoting Efficient Water Use in Agro-based Industries, TERI Press, New Delhi, India, 2004, pp. 167–169.
- [72] N. Gulati, Conservation of resources using evaporation and spray drying technology for distillery and paper industries, in: P.K. Tewari (Ed.), Liquid Asset. Proceeding of Indo-EU Workshop on Promoting Efficient Water Use in Agro-based Industries, TERI Press, New Delhi, India, 2004, pp. 163–166.
- [74] W.J.B.M. Driessen, M.H. Tielbaard, T.L.F., Vereije, Experience on anaerobic treatment of distillery effluer with the uasb process, Water Sci. Technol. 30 (1994) 193–200.
- [75] F.J. Fitzgibbon, P. Nigam, D. Sichh, R. Kurshant, Biological treatment of distillery waste for polluting remediation, "Basic M⁴ ob. 35 (1995) 293–302.
- [76] K.S. Jayantha, J.K. Ramaujan, Steep of criteria to a upflow anaerobic sludge blanket (UASB) record, Physical Rev and States and States
- [77] K. Kida, S. Moritte, N. Abes, a gonoda, Birthgical treatment of schochu distillery waster ter, Process. Birchnolm dv. 30 (29) (1995) 125–132.
- [78] R. Rajor, Prosing, R.P. Mathur, removal of distillery waste by Sacchronices, Indian Environ. Prot. 22 (12) (2000) 1241–1252.
- [79] S. Jain, M. Balakrishn, Membrane fouling and cleaning in ultrafiltratic or a derobically treate distillery effluent, in: Euro-membrane 2004, amburg, Germany, 28 September–1 October, 2004.